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The n:p ratio for Cr-53 is $\frac{29}{24} = 1.21$; for Mn-51, it is $\frac{26}{25} = 1.04$; for Fe-59, it is $\frac{33}{26} = 1.27$. Positron decay occurs when the n:p ratio is low. Positron decay occurs when the n:p ratio is low.

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and fusion

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CHAPTER 21 REVIEW Nuclear Chemistry

Chemistry Predict the mode of decay of (a) carbon-14, (b) (b) xenon-118. (b) Xenon has an atomic number of 54. Thus, xenon-118 has 54 protons and $118 - 54 = 64$ neutrons, giving it a neutron-to-proton ratio of According to Figure 21.2 , stable nuclei in this region of the belt

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Section 21.1 • Introduction to Hydrocarbons 745 Figure 21.3 Methane—a hydrocarbon found in natural gas—is the simplest hydrocarbon. Identify In addition to hydrogen, what other elements readily bond with carbon? Organic chemistry Today, the term organic compound is applied to all carbon-containing compounds with the primary exceptions of

Chapter 21: Hydrocarbons

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(a) (b) Answer is one of the following. A/yellow: mass = 65.14 kg, volume = 3.38 L, density = 19.3 kg/L, likely identity = gold. B/blue: mass = 0.64 kg, volume = 1.00 L, density = 0.64 kg/L, likely identity = apple.

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